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SWIMMERS PUT PENN TO STIFF TEST

INSTITUTE ATHLETES HAVE UNSUCCESSFUL WEEK-END

(Continued from Page 1)

off the reel, and ended only when at the end of the fifth minute, the Technology man secured a body-scissors and half nelson on his opponent, and so the victory.

The decision rendered in the 135 pound bout gave Yale her first tally. In this bout, Moore of the Elis obtained the decision over Hazeltine of the Engineers.

Captains Battle on Mat

It remained for the rival captains, Mallon of Yale and Butler of M. I. T. to furnish the numerous spectators with the feature bout of the evening. Both contestants evinced their ability at the wrestling game by very clever mat work, and the nine-minute bout was replete with action. In the extra periods awarded, Butler obtained the decision.

Benjamin of Yale furnished his "Alma Mater" with her second and final tally by obtaining a decision over Briggs of the Institute. The visiting grappler was never in much danger, as his opponent always endeavored to play on the defensive.

In the unlimited class bout, Humphrey took the mat for M. I. T., against McKay of Yale. Humphrey, although himself of no slight stature, seemed a pigmy beside his foot-ball opponent. The Eli foot-ball guard tried the tactics that gave him the victory at Columbia, but the Beaver was too wary. The period ended with neither of the contestants coming to determinate grips. In the extra period, Humphrey obtained his favorite body-scissors, and kept his advantage almost to the end of the bout, and so obtained the decision.

Ducks Put Penn to Stiff Test

Contrary to the expectation of members of the team, and of a number of the spectators, Pennsylvania took a 28-25 victory Saturday night at the Y. M. C. A. tank. The meet showed some close events, and the result

was undecided up to the last seconds of the final event.

Plenty of Thrills

At the start of the century the score was 19-25 and it was up to the Beavers to take the first and second places in the event to win the meet. Captain Biddell came through with first, outstripping the speedy Rambr, and Trowbridge showed some good work against the Keystone star. In the last lap however, Rambr forged ahead, and took the coveted second place.

In the 220 Doane Greene put up a mighty hard fight against Rambr, and touched the finish line only a fraction of a second after the visitor. The 50 was nearly as close, when Weiner defeated Palmer. Both teams were working hard, and no event was won by a wide margin, except the plunge, in which Bursh made over 70 feet for Penn.

One Milers Win in Fast Time

The one-mile relay team soundly whipped the runners representing Penn State at the Johns Hopkins Relay carnival at Baltimore Saturday night, breasting the tape about 35 yards in the van of the Staters.

The team went to Baltimore with the idea that they were billed to run Pennsylvania, one mile champions of America, but due to a mistake in concluding the arrangements, when they arrived in Baltimore they found they were up against Penn State.

Frankie Howlett, Bill Gurney, Olie Bardes, and Yardley Chittick running in the order named represented Technology and they turned in the fastest time of the night, better by three seconds than any other mile relay team did that evening and

there was a host of them running, too.

Frankie Howlett led off for Technology and found in Moore of Penn State an opponent just as fast as he was dusky and though Howlett ran a good race Moore was able to hand over a very slight lead to his second man, while Bill Gurney got away just that much behind. But Bill was not long in making up the distance and he gave his man a bad beating. Bardes and Chittick both increased the Technology lead till the teams were separated by 35 yards at the tape.

Dartmouth Gym Team Cops Victory

The Dartmouth gym squad left Technology in the lurch, 31-23, last Saturday night as the closing event

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What Is Air Pressure?

THE air is composed of molecules. They constantly bombard you from all sides. A thousand taps by a thousand knuckles will close a barn door. The taps as a whole constitute a push. So the constant bombardment of the air molecules constitutes a push. At sea-level the air molecules push against every square inch of you with a total pressure of nearly fifteen pounds.

Pressure, then, is merely a matter of bombarding molecules.

When you boil water you make its molecules fly off. The water molecules collide with the air molecules. It takes a higher temperature to boil water at sea-level than on Pike's Peak. Why? Because there are more bombarding molecules at sea-level—more pressure.

Take away all the air pressure and you have a perfect vacuum. A perfect vacuum has never been created. In the best vacuum obtainable there are still over two billion molecules of air per cubic centimeter, or about as many as there are people on the whole earth.

Heat a substance in a vacuum and you may discover properties not revealed under ordinary pressure. A new field for scientific exploration is opened.

Into this field the Research Laboratories of the General Electric Company have penetrated. Thus one of the chemists in the Research Laboratories studied the disintegration of heated metals in highly exhausted bulbs. What happened to the glowing filament of a lamp, for example? The glass blackened. But why? He discovered that the metal distilled in the vacuum depositing on the glass.

This was research in pure science—research in what may be called the chemistry and physics of high vacua. It was undertaken to answer a question. It ended in the discovery of a method of filling lamp bulbs with an inert gas under pressure so that the filament would not evaporate so readily. Thus the efficient gas-filled lamp of today grew out of a purely scientific inquiry.

So, unforeseen, practical benefits often result when research is broadly applied.

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